

## SYNTHESIS OF 4,5-SECO-DERIVATIVES OF ALLOBETULIN

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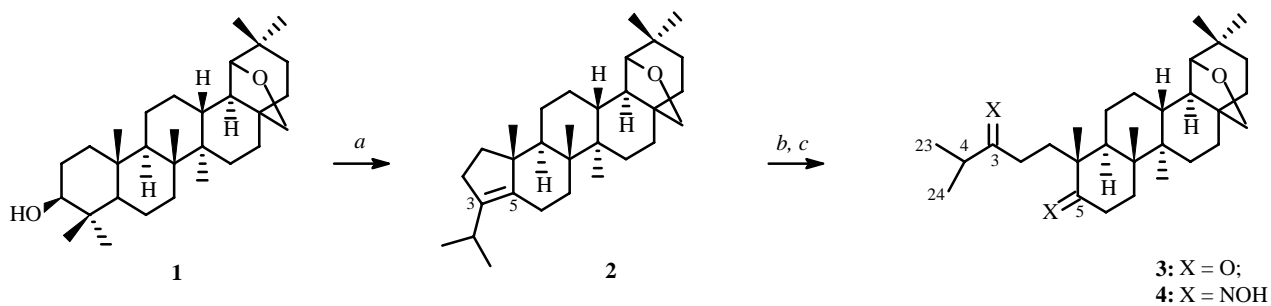
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*4,5-Seco-19 $\beta$ ,28-epoxy-18 $\alpha$ -olean-3,5-dione was prepared by ozonation of 19 $\beta$ ,28-epoxy-18 $\alpha$ -olean-3(5)-ene. The structures of the synthesized compounds were confirmed using spectral data.*

**Key words:** oleanane triterpenoids, seco-derivatives, ozonolysis.

Allobetulin is a pentacyclic oleanane triterpenoid found among triterpene compounds of birch bark. It is easily formed from betulin via acid-catalyzed isomerization. It has been found that allobetulin possesses moderate inhibitory activity for type B flu virus [1] and that 28-oxoallobetulone is an effective inhibitor of type A flu virus multiplication [2]. The *nor*-derivative of allobetulin is an antifeedant for *Heliothis zea* larvae [3]. Derivatives of allobetulin are used as biomarkers [4]. 3,4-Seco-derivatives of allobetulin possess antibacterial activity [5]. Furthermore, highly active inhibitors of nitric oxide production in murine macrophages were found among oleanane triterpenoids with a modified ring A [6-8]. 19 $\beta$ ,28-Epoxy-18 $\alpha$ -oleanane 1,2-seco-lactone was synthesized by oxidation of 3-keto-allobetulin with CrO<sub>3</sub> in AcOH [9]. 3,4-Seco-derivatives of allobetulin can be prepared by decomposing lactones, Baeyer—Villiger oxidation products of triterpene 3-ketones [10, 11]. Nevertheless, the synthesis of 4,5-seco-oleanane triterpenoids has not been reported.

In order to synthesize 4,5-seco-derivatives of allobetulin, we developed a scheme based on ozonolytic cleavage of the *endo*-cyclic  $\Delta^{3,5}$ -bond of 19 $\beta$ ,28-epoxy-18 $\alpha$ -olean-3(5)-ene (**2**), which, in turn, was prepared from allobetulin (**1**) by Wagner—Meerwein rearrangement in the presence of PCl<sub>5</sub> in a benzene:toluene mixture at room temperature in 80% yield. The structures of the prepared compounds were established using NMR spectroscopy. Thus, the <sup>13</sup>C NMR spectrum of **2** contains signals for C-3 and C-5 ( $\delta$  139.8 and 136.1 ppm); the PMR spectrum, a signal for H-4 at  $\delta$  2.65 ppm as a septet ( $J = 6.8$  Hz).



*a.* PCl<sub>5</sub>, benzene:toluene, 20°C; *b.* O<sub>3</sub>, -60 °C, Zn/AcOH; *c.* NH<sub>2</sub>OH·HCl, pyridine

Ozonation of **2** in CH<sub>2</sub>Cl<sub>2</sub> at -60°C gave 4,5-seco-19 $\beta$ ,28-epoxy-18 $\alpha$ -olean-3,5-dione (**3**). The yield after chromatographic purification was 56%. The <sup>13</sup>C NMR spectrum contains signals for the C-3 and C-5 ketones ( $\delta$  214.4 and 217.0 ppm); the PMR spectrum, a signal for H-4 as a septet at  $\delta$  2.55 ppm. Signals for the C-23 and C-24 methyls resonate at  $\delta$  ~1.00 ppm as characteristic doublets. The structure of the 4,5-seco-fragment was confirmed by data for 4,5-seco-19 $\beta$ ,28-epoxy-18 $\alpha$ -olean-3,5-dioxime (**4**), prepared by reaction of triterpene 3,5-diketone **3** with hydroxylamine. Thus, the <sup>13</sup>C NMR

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spectrum of **4** has signals for C-3 and C-5 shifted to strong field compared with those of diketone **3** ( $\delta$  167.5 and 167.9 ppm). The signal for H-4 in the PMR spectrum remained as a septet. Signals for H-2 and H-6 were observed at 2.10-2.45 ppm.

Thus, we prepared for the first time 4,5-seco-derivatives of allobetulin that are interesting as potential biologically active compounds.

## EXPERIMENTAL

IR spectra were recorded on Specord M80 spectrometers for samples as mulls in mineral oil. PMR and  $^{13}\text{C}$  NMR spectra were recorded on an AM-300 spectrometer (Bruker, 300 and 75.5 MHz, respectively) in  $\text{CDCl}_3$  with TMS internal standard. Melting points were determined on a Boetius microstage. Optical density was measured on a 241 MC polarimeter (Perkin—Elmer) in a 1-dm tube. TLC was performed on Silufol plates (Chemapol, Czech. Rep.) using  $\text{CHCl}_3:\text{CH}_3\text{OH}$  (25:1). Compounds were developed with phosphotungstic acid in ethanol (20%) with subsequent heating at 100-120°C for 2-3 min. Allobetulin (**1**) was prepared as before [12].

Elemental analyses of all compounds agreed with those calculated.

**19 $\beta$ ,28-Epoxy-18 $\alpha$ -olean-3(5)-ene (2).** A solution of **1** (1 mmol, 0.44 g) in anhydrous benzene:toluene (100 mL, 1:1) at 20°C was treated in one portion with  $\text{PCl}_5$  (5.5 mmol, 0.50 g), stirred for 30 min (TLC monitoring), diluted with saturated aqueous  $\text{Na}_2\text{CO}_3$  (30 mL), stirred for another 30 min, and brought to room temperature. The organic layer was separated, washed with water ( $3 \times 20$  mL), dried over  $\text{Na}_2\text{SO}_4$ , and evaporated in vacuum. The solid was crystallized from ethanol. Yellow compound,  $\text{C}_{30}\text{H}_{48}\text{O}$ , yield 0.34 g (80%),  $R_f$  0.85, mp 199-201°C,  $[\alpha]_D^{20} +82^\circ$  ( $c$  1.0,  $\text{CHCl}_3$ ), lit. mp [13] 200-201°C,  $[\alpha]_D^{22} +81^\circ$  ( $c$  1.23). IR spectrum ( $\nu$ ,  $\text{cm}^{-1}$ ): 1760, 1620, 1530, 1460, 1360, 1310, 1280, 1240, 1160, 1040, 910, 790, 720.

PMR spectrum ( $\text{CDCl}_3$ ,  $\delta$ , ppm, J/Hz): 0.79, 0.85, 0.86, 0.92, 0.98, 1.00, 1.03 (7s, 21H, 7 $\text{CH}_3$ ), 1.10-2.30 (m, 23H,  $\text{CH}_2$ , CH), 2.65 (1H, septet, H-4,  $J = 6.8$ ), 3.42 and 3.80 (both d, 1H each, H-28,  $J = 7.8$ ), 3.54 (s, 1H, H-19).

$^{13}\text{C}$  NMR spectrum ( $\text{CDCl}_3$ ,  $\delta$ , ppm): 13.4, 14.2, 19.0, 19.7, 21.3, 21.8, 23.6, 24.5, 26.2, 26.3, 26.5, 26.7, 27.3, 28.7, 32.4, 32.6, 34.5, 36.2, 36.7, 40.6, 40.8, 41.4, 42.1, 46.7, 49.8, 50.0, 71.2 (C28), 87.9 (C19), 136.1 (C4), 139.8 (C3).

**4,5-Seco-19 $\beta$ ,28-epoxy-18 $\alpha$ -olean-3,5-dione (3).** Ozone was passed through a solution of **2** (2 mmol, 0.85 g) in  $\text{CH}_2\text{Cl}_2$  (50 mL) at -60°C until the starting material disappeared (TLC monitoring). The temperature was raised to 0°C. Glacial AcOH (10 mL) and zinc dust (1 g) were added. The mixture was stirred for 1 h and filtered. The organic layer was washed with saturated  $\text{Na}_2\text{CO}_3$  solution ( $2 \times 20$  mL) and water ( $2 \times 20$  mL), dried over  $\text{Na}_2\text{SO}_4$ , and evaporated in vacuum. The solid was purified by column chromatography over  $\text{Al}_2\text{O}_3$  with elution by benzene. Yellow compound,  $\text{C}_{30}\text{H}_{48}\text{O}_3$ , yield 0.51 g (56%),  $R_f$  0.40, mp 238-239°C,  $[\alpha]_D^{20} +39^\circ$  ( $c$  1.00,  $\text{CHCl}_3$ ). IR spectrum ( $\nu$ ,  $\text{cm}^{-1}$ ): 1720, 1470, 1380, 1160, 1120, 1070, 1040, 1020, 980, 950, 900, 870, 850, 780, 730.

PMR spectrum ( $\text{CDCl}_3$ ,  $\delta$ , ppm, J/Hz): 0.75, 0.87, 0.88, 0.94, 0.97 (15H, 5s, 5 $\text{CH}_3$ ), 1.01 and 1.02 (3H each, both d,  $\text{CH}_3$ -23,  $J = 1.5$ ,  $\text{CH}_3$ -24,  $J = 1.4$ ), 1.10-1.90 (19H, m,  $\text{CH}_2$ , CH), 2.02-2.08 (2H, m, H-2), 2.31-2.39 (2H, m, H-6), 2.55 (1H, septet, H-4,  $J = 6.7$ ), 3.40 and 3.78 (1H each, both d, H-28,  $J = 7.8$ ), 3.54 (1H, s, H-19).

$^{13}\text{C}$  NMR spectrum ( $\delta$ , ppm): 13.3, 15.5, 18.1, 18.3, 20.3, 23.0, 24.4, 25.9, 26.8, 28.6, 30.0, 32.5, 33.4, 34.3, 35.0, 35.6, 36.2, 36.5, 38.9, 40.5, 40.8, 41.3, 4.15, 44.7, 46.4, 49.6, 71.0 (C-28), 87.7 (C-19), 214.4 and 217.0 (C-3, C-5).

**4,5-Seco-19 $\beta$ ,28-epoxy-18 $\alpha$ -olean-3,5-dioxime (4).** A solution of **3** (1 mmol, 0.46 g) in anhydrous pyridine (20 mL) was treated with  $\text{NH}_2\text{OH}\cdot\text{HCl}$  (1 g), boiled for 2 h, and poured into HCl (50 mL, 5%). The precipitate was filtered off, washed with water, and dried. Yellow compound,  $\text{C}_{30}\text{H}_{50}\text{N}_2\text{O}_3$ , yield 0.37 g (80%),  $R_f$  0.30, mp 143-145°C,  $[\alpha]_D^{20} +89^\circ$  ( $c$  1.00,  $\text{CHCl}_3$ ). IR spectrum ( $\nu$ ,  $\text{cm}^{-1}$ ): 1710, 1670, 1600, 1460, 1390, 1320, 1300, 1280, 1260, 1170, 1150, 1110, 1040, 940, 720.

PMR spectrum ( $\text{CDCl}_3$ ,  $\delta$ , ppm, J/Hz): 0.79, 0.87, 0.90, 0.94, 0.99 (15H, 5s, 5 $\text{CH}_3$ ), 1.13 (6H, s,  $\text{CH}_3$ -23,  $\text{CH}_3$ -24), 1.00-1.90 (19H, m,  $\text{CH}_2$ , CH), 2.10-2.18 (2H, m, H-2), 2.35-2.45 (2H, m, H-6), 2.62 (1H, septet, H-4,  $J = 6.7$ ), 3.41 and 3.80 (1H each, both d, H-28,  $J = 7.8$ ), 3.54 (1H, s, H-19), 9.20 (2H, m, 2 $\text{NOH}$ ).

$^{13}\text{C}$  NMR spectrum ( $\delta$ , ppm): 13.3, 14.8, 15.1, 15.6, 19.3, 19.9, 20.1, 23.7, 24.5, 26.0, 26.2, 26.3, 26.8, 28.7, 32.6, 33.7, 34.6, 36.2, 36.6, 37.3, 39.4, 40.5, 41.5, 42.5, 42.7, 46.6, 71.2 (C-28), 87.8 (C-19), 164.6 and 165.8 (C-5, C-3).

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## REFERENCES

1. V. G. Platanov, A. D. Zorina, M. A. Gordon, N. P. Chizhov, L. V. Balykina, Yu. D. Mikhailov, D. R. Ivanen, Tran Kim Kvi, and A. G. Shawa, *Khim.-Farm. Zh.*, **2**, 42 (1995).
2. O. B. Flekhter, L. R. Nigmatullina, L. A. Baltina, L. T. Karachurina, F. Z. Galin, F. S. Zarudii, G. A. Tolstikov, E. I. Boreko, N. I. Pavlova, S. N. Nikolaeva, and O. V. Savinova, *Khim.-Farm. Zh.*, **36**, No. 9, 26 (2002).
3. F. N. Lugemwa, F. Y. Huang, M. D. Bentley, M. J. Mendel, and A. R. Alford, *J. Agric. Food Chem.*, **38**, 493 (1990).
4. A. White, E. J. Horsington, N. Nedjar, N. Peakman, and J. A. Curiale, *Tetrahedron Lett.*, **39**, 3931 (1998).
5. I. Valterova, J. Klinot, and A. Vystrcil, *Collect. Czech. Chem. Commun.*, **48**, 649 (1983).
6. T. Honda, B. A. V. Rounds, L. Bore, H. J. Finlay, F. G. Favaloro, G. W. Gribble, N. Suh, Y. Wang, and M. B. Sporn, *Bioorg. Med. Chem. Lett.*, **9**, 3429 (1999).
7. T. Honda, B. A. V. Rounds, L. Bore, H. J. Finlay, F. G. Favaloro, N. Suh, Y. Wang, M. B. Sporn, and G. W. Gribble, *J. Med. Chem.*, **43**, No. 22, 4233 (2000).
8. T. Honda, G. W. Gribble, N. Suh, H. J. Finlay, B. Rounds, L. Bore, F. G. Favaloro, Y. Wang, and M. B. Sporn, *J. Med. Chem.*, **43**, No. 9, 1866 (2000).
9. J. Sejbal, M. Homolova, I. Tislerova, and V. Krecek, *Collect. Czech. Chem. Commun.*, **65**, 1339 (2000).
10. J. Sejbal, J. Klinot, D. Hrcirova, and A. Vystrcil, *Collect. Czech. Chem. Commun.*, **50**, 2753 (1985).
11. J. Sejbal, J. Klinot, and A. Vystrcil, *Collect. Czech. Chem. Commun.*, **52**, 1052 (1987).
12. T.-S. Li, J.-X. Wang, and X.-J. Zheng, *J. Chem. Soc. Perkin Trans. I*, 3957 (1998).
13. G. R. Pettit, B. Green, and W. J. Bowyer, *J. Org. Chem.*, **26**, 2879 (1961).